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table of  
ionization

constants and  $K_a$ 's is required  
to work some of  
the problems in

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In which one of  
the following  
solutions will  
acetic acid have  
the greatest  
percent  
ionization?

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Properties Of

Solutions

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Answers can be solids, liquids, or gases.

Remember: The solvent

dissolves the solute. Three factors that determine the

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Answers

rate at which a  
solute dissolves  
are stirring  
(agitation),  
temperature, and  
the particle  
size of the  
solute. Chapter  
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After achieving

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will provide you

cheat for PP

Crossword

Chapter 16

answers, this

game was

developed by

AppyNation Ltd.

the famous

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developer known  
in word puzzle  
game. The game  
has 16 Chapters,  
each one  
contains many  
crosswords built  
with pictures.  
Many answers are  
easy to find but  
everybody could  
stuck in one or  
another pic.

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*PP Crossword*

*Chapter 16*

*Answers and*

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Properties Of

Solutions

Answers In all

solutions,

whether gaseous,

liquid, or

solid, the

substance

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Solutions in the  
greatest amount  
is the solvent,  
and the  
substance or  
substances  
present in  
lesser amounts  
are the solute  
(s). The solute  
does not have to  
be in the same  
physical state  
as the solvent,



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*Solutions*

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Properties Of

Solutions

Worksheet

Answers can be

solids, liquids,

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Solutions

Remember: The  
solvent

dissolves the  
solute. Three  
factors that  
determine the  
rate at which a  
solute dissolves  
are stirring  
(agitation),  
temperature, and  
the particle  
size of the

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SOLUTIONS (pages  
471-477) This  
section

identifies the  
factors that  
affect the  
solubility of a  
substance and  
determine the  
rate at which a  
solute  
dissolves.

Solution

Formation (pages

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471-472) Look at  
Figure 16.1 on  
page 471 to ...  
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Answers Chapter  
16 Page 7/31

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means to

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options to  
accompany you  
like having  
additional time.  
It will not  
waste your time.  
receive me, the  
e-book will  
unconditionally  
declare Page 2/9

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*Solutions*

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Bases Worksheet

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(Section 16.5,

*Page 32/51*



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Solutions 16.8) to  
complete this  
worksheet. 1.

Answers  
Select the  
lightbulb and  
drag the  
electrodes into  
the liquid. a.  
Which solutions  
produce a bright  
light? b. Which  
solutions  
produce a dim  
light? 2.

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16, Solutions

(continued) 7.

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Look at Figure 16.2 on page 472. Circle the letter of each sentence that is true about a saturated solution.

a. The total amount of dissolved solute remains constant.

b. The total mass of undissolved

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crystals remains  
constant. c.

When the rate of  
solvation equals  
the

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16 - 1 Mark

*Page 38/51*

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Solutions and  
Answers.

Question 1.

Define

dispersion of  
light. [MSE

(Chandigarh)

2007] Answer:

Splitting up of  
white light into  
seven colours  
when it passes  
through a glass  
prism is known

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Answers

as dispersion of light. Question 2. Name the colours in the order they appear in the spectrum of light.

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If you intention  
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Properties Of  
Solutions

Answers In all  
solutions,  
whether gaseous,  
liquid, or  
solid, the  
substance  
present in the

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Solutions  
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Answers

greatest amount  
is the solvent,  
and the  
substance or  
substances  
present in  
lesser amounts  
are the solute  
(s). The solute  
does not have to  
be in the same  
physical state  
as the solvent,  
but

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PROPERTIES OF  
SOLUTIONS (pages  
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section

identifies the

factors that

affect the

solubility of a

substance and

determine the

rate at which a

solute

dissolves.



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Solution

Formation (pages  
471-472) Look at

Figure 16.1 on  
page 471 to help  
you answer

Questions 1 and  
2. 1.

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for Class 5 EVS  
Chapter 16 Who

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Will Do This  
Work? QUESTIONS  
FROM TEXTBOOK  
SOLVED 1. Have  
you seen such  
scenes around  
you? Ans. Yes, I  
have seen such  
scenes around  
me. 2. Have you  
ever thought of  
people who do  
this work? Can  
you imagine how

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they would feel?

Ans. Doing such  
works must be  
[...]

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