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Some chemical elements have several isotopes. Some of those are unstable and fall apart, releasing radiation. We've figured out how fast every kind of isotope falls apart. they all have a "half-life", the time in which half the atoms of an isotope decay and in which the intensity of the radiation is halved.

How is radiometric dating used to ... - Yahoo Answers

1. After three half-lives of an isotope, 1 billion (or 1/8) of the original isotope's atoms remain. How many atoms of the daughter product would you expect to be present? 2. By measuring the amounts of parent isotope and daughter product in the minerals contained in a rock, and by knowing the half-life of the parent isotope, a geologist can calculate the absolute age of a rock. A rock contains ...

Radiometric Dating - Isotope, Atoms...? | Yahoo Answers

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Radiometric dating measures the decay of radioactive atoms to determine the age of a rock sample. It is founded on unprovable assumptions such as 1) there has been no contamination and 2) the decay rate has remained constant.

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Featured in Answers Magazine Radiometric dating is often used to “prove” rocks are millions of years old. Once you understand the basic science, however, you can see how wrong assumptions lead to incorrect dates. Most people think that radioactive dating has proven the earth is billions of

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years old.

Radiometric Dating: Back to Basics | Answers in Genesis
Solution for Radiometric dating is a technique used to determine the age of rocks. Geologists have used this technique to determine that:

Answered: Radiometric dating is a technique used... | bartleby
Radiometric dating doesn't work for telling us how old the pyramids are for this reason. There are other techniques that rely on radiation, but not on radioactive decay. Luminescence dating is a technique used by archeologists and geologists which determines how long it has been since something was last exposed to sunlight or heat.

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